# Valorizing Carbon Dioxide into Commodity Chemicals (& Energy Carriers) using Catalysis





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### What are Catalysts?



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#### From web image search

#### From AI (Imagen 2)

### What are Catalysts?

#### An alternate analogy

Catalysts are molecular machines...

...they assemble feedstock parts



After constructing product, they start the same process again

### **Motivations and Inspirations**

### CO<sub>2</sub>-A Renewable C<sub>1</sub> Feedstock

- Highly abundant (10<sup>14</sup> tons in atmosphere and oceans)
- Cheap availability
- Essentially non-toxic (compared to Cl<sub>2</sub>C=O & CO)
- It sticks to metals (the active part of most catalysts) reasonably well



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**Commodity Chemicals** 

chemicals we use to make other stuff Molecular Energy Sources chemicals we use as fuels or to store fuels

## **Current & Recent CO<sub>2</sub> Catalysis Projects and \$**

#### Sustainable Polymer Building Blocks



## **Current & Recent CO<sub>2</sub> Catalysis Projects and \$**





## **Reversible CO<sub>2</sub> Hydrogenation**

For Chemicals & Energy



### If At First You Fail...



Chem. Sci. 2015, 6, 4291.

ACS Catal. 2015, 5, 2404.

### ...Add Lewis Acid

#### Formic acid dehydrogenation



2° PNP Fe(II) w/o LA: 200 TON 2° PNP Fe(II) w/ LA: ~1 x 10<sup>6</sup> TON

J. Am. Chem. Soc. **2014**, 136, 10234. CO<sub>2</sub> hydrogenation to formate



2° PNP Fe(II) w/o LA: 900 TON 3° PNP Fe(II) w/o LA: 1100 TON **3° PNP Fe(II) w/ LA: 46,000TON** 

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#### Methanol dehydrogenation



2° PNP Fe(II) w/o LA: 350 TON 2° PNP Fe(II) w/ LA: 51,000 TON

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### Lewis Acid Influence Is Widespread



### So What is the Lewis Acid Doing?

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xs base

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### **Mechanisms of Hydrogenation Catalysis**



143, 10631.